# How honey kills bacteria

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ABSTRACT With the rise in prevalence of antibioticresistant bacteria, honey is increasingly valued for its antibacterial activity. To characterize all bactericidal factors in a medical-grade honey, we used a novel approach of successive neutralization of individual honey bactericidal factors. All bacteria tested, including Bacillus subtilis, methicillin-resistant Staphylococcus aureus, extended-spectrum β-lactamase producing Escherichia coli, ciprofloxacin-resistant Pseudomonas aeruginosa, and vancomycin-resistant Enterococcus faecium, were killed by 10-20% (v/v) honey, whereas  $\geq 40\%$ (v/v) of a honey-equivalent sugar solution was required for similar activity. Honey accumulated up to  $5.62 \pm$  $0.54 \text{ mM H}_{2}\text{O}_{2}$  and contained  $0.25 \pm 0.01 \text{ mM}$  methylglyoxal (MGO). After enzymatic neutralization of these two compounds, honey retained substantial activity. Using B. subtilis for activity-guided isolation of the additional antimicrobial factors, we discovered bee defensin-1 in honey. After combined neutralization of H<sub>2</sub>O<sub>2</sub>, MGO, and bee defensin-1, 20% honey had only minimal activity left, and subsequent adjustment of the pH of this honey from 3.3 to 7.0 reduced the activity to that of sugar alone. Activity against all other bacteria tested depended on sugar, H<sub>2</sub>O<sub>2</sub>, MGO, and bee defensin-1. Thus, we fully characterized the antibacterial activity of medical-grade honey.—Kwakman, P. H. S., te Velde, A. A., de Boer, L., Speijer, D., Vandenbroucke-Grauls, C. M. J. E., Zaat, S. A. J. How honey kills bacteria. FASEB J. 24, 2576-2582 (2010). www.fasebj.org

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Honey has been renowned for its wound-healing properties since ancient times (1). At least part of its positive influence is attributed to antibacterial properties (2, 3). With the advent of antibiotics, clinical application of honey was abandoned in modern Western medicine, although in many cultures, it is still used (4). These days, however, abundant use of antibiotics has resulted in widespread resistance. With the development of novel antibiotics lagging behind (5), alternative antimicrobial strategies are urgently needed.

The potent *in vitro* activity of honey against antibiotic-resistant bacteria (6, 7) and its successful application in treatment of chronic wound infections not responding to antibiotic therapy (3) have attracted considerable attention (8-10).

The broad spectrum antibacterial activity of honey is multifactorial in nature. Hydrogen peroxide and high osmolarity—honey consists of  $\sim\!80\%$  (w/v) of sugars—are the only well-characterized antibacterial factors in honey (11). Recently, high concentrations of the antibacterial compound methylglyoxal (MGO) were found specifically in Manuka honey, derived from the Manuka tree (*Leptospermum scoparium*) (12, 13). Until now, no honey has ever been fully characterized, which hampers clinical application of honey.

Recently, we determined that Revamil medical-grade honey, produced under standardized conditions in greenhouses, has potent, reproducible bactericidal activity (14). In the current study, we identified all bactericidal factors in the honey used as source for this product and assessed their contribution to honey bactericidal activity.

To accomplish this, we used a novel approach of successive neutralization of individual honey bactericidal factors combined with activity-guided identification of unknown factors.

## MATERIALS AND METHODS

#### Honey

Unprocessed Revamil source (RS) honey was kindly provided by Bfactory Health Products (Rhenen, The Netherlands). RS honey has a density of 1.4 kg/L and contains 333 g/kg glucose, 385 g/kg fructose, 73 g/kg sucrose, and 62 g/kg maltose. To study the contribution of the sugars to the bactericidal activity of honey, a solution with a sugar composition identical to that of the honey was prepared.

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### Microorganisms

Bactericidal activity of honey was assessed against the laboratory strains *Bacillus subtilis* ATCC6633, *Staphylococcus aureus* 42D, *Escherichia coli* ML-35p (15), and *Pseudomonas aeruginosa* PAO-1 (ATCC 15692), and against clinical isolates of methicillin-resistant *S. aureus* (MRSA), vancomycin-resistant *Enterococcus faecium* (VREF), extended-spectrum β-lactamase-producing *E. coli* (*E. coli* ESBL) and ciprofloxacin-resistant *P. aeruginosa* (CRPA).

#### Determination of H<sub>2</sub>O<sub>2</sub> concentration in honey

Hydrogen peroxide concentrations in honey were determined quantitatively using a modification of a method described previously (16). Undiluted and 10-fold diluted samples of honey (40 μl) were mixed in wells of microtiter plates with 135 μl reagent, consisting of 50 μg/ml O-dianisidine (Sigma, St. Louis, MO, USA) and 20 μg/ml horseradish peroxidase type IV (Sigma) in 10 mM phosphate buffer (pH 6.5). O-dianisidine and peroxidase solutions were freshly prepared from a 1 mg/ml stock in demineralized water and from a 10 mg/ml stock in 10 mM phosphate buffer (pH 6.5), respectively. After 5-min incubations at room temperature, reactions were stopped by addition of 120 μl 6 M  $\rm H_2SO_4$ , and absorption at 540 nm was measured. Hydrogen peroxide concentrations were calculated using a calibration curve of 2-fold serial dilutions of  $\rm H_2O_2$  ranging from 2200 to 2.1 μM.

#### MGO neutralization assay

Reduced glutathione (Sigma) was added to diluted honey to a final concentration of 15 mM, and conversion of MGO to S-D-lactoyl-glutathione (SLG) was initiated by addition of 0.5 U/ml glyoxalase I (Sigma). The amount of MGO converted was determined using the extinction coefficient of SLG of  $3.37~\rm mM^{-1}$  at 240 nm (17). Thus, we determined that up to 10 mM of exogenous MGO added to 40% honey was completely converted, and that undiluted RS honey contained  $0.25 \pm 0.01~\rm mM$  of MGO.

#### Antibee defensin-1 polyclonal antibody

An affinity-purified polyclonal antibee defensin-1 antibody was purchased from Eurogentec (Seraing, Belgium). The N-terminal part of bee defensin-1 is hydrophobic and contains 3 disulfide bonds, whereas the hydrophilic C-terminal region lacks cysteine residues (18). Therefore, rabbits were immunized with a synthetic peptide corresponding to the C terminus of bee defensin-1 (CRKTSFKDLWDKRF), and antibodies were subsequently affinity-purified using this peptide coupled to AF-Amino Toyopearl 650 M resin (Toso, Tokyo, Japan).

#### Liquid bactericidal assay

Bactericidal activity of honey was quantified in 100- $\mu$ l volume liquid tests, in polypropylene microtiter plates (Costar Corning, New York, NY, USA). For each experiment, a 50% (v/v) stock solution of honey was freshly prepared in incubation buffer containing 10 mM phosphate buffer (pH 7.0) supplemented with 0.03% (w/v) trypticase soy broth (TSB; BD Difco, Detroit, MI, USA). Bacteria from logarithmic phase cultures in TSB were washed twice with incubation buffer and suspended at a final concentration of  $1 \times 10^6$  CFU/ml, based on optical density. Plates were incubated at 37°C on a rotary shaker at 150 rpm. At indicated time points, duplicate 10- $\mu$ l

aliquots of undiluted and 10-fold serially diluted incubations were plated on blood agar. Bacterial survival was quantified after overnight incubation at  $37^{\circ}$ C. The detection level of this assay is 100~CFU/ml.

To assess the contribution of  $\rm H_2O_2$  to the bactericidal activity of honey, bovine liver catalase (Sigma) was added to a final concentration of 600 U/ml. A catalase stock solution was prepared according to the manufacturers' instructions in 50 mM phosphate buffer (pH 7.0). The addition of 0.25% (v/v) of this catalase stock solution reduced the amount of  $\rm H_2O_2$  to undetectable levels at all honey concentrations tested and did not affect bacterial viability.

Sodium polyanetholsulfonate (SPS) (Sigma) was added to neutralize cationic bactericidal components (19) at a final concentration of 0.025% (w/v). The incubation buffer did not affect the pH of the concentrations of honey used in our experiments. A 1 M NaOH solution was used to titrate honey solutions to pH 7.0.

#### Agar diffusion assay

To assess antibacterial activity of fractionated honey, an agar diffusion assay was used (20). In brief, a *B. subtilis* inoculum suspension was prepared as described for the liquid bactericidal assay. Bacteria (10<sup>7</sup> CFU) were mixed with 20 ml nutrient-poor agar [0.03% (w/v) TSB in 10 mM sodium phosphate buffer (pH 7.0) with 1% low EEO agarose (Sigma)] of 45°C, and immediately poured into 10- × 10-cm culture plates. Wells of 1 mm diameter were punched into the agarose, and 2.5-µl samples were added to the wells and allowed to diffuse into the agarose for 3 h at 37°C. Subsequently, the agarose was overlaid with 20 ml of double-strength nutrient agarose [6% TSB and 1% Bacto-agar (BD Difco), 45°C], and plates were incubated overnight at 37°C. Clear zones around the wells indicated antibacterial activity.

#### Ultrafiltration of honey components

Fifteen milliliters of 20% honey was centrifuged in a 5-kDa molecular weight cutoff Amicon Ultra-15 tube (Millipore, Bedford, MA, USA) at 4000 g for 45 min at room temperature. The <5-kDa filtrate was collected, and the >5-kDa retentate was subsequently washed 3 times in the filter tube with 15 ml of demineralized water and concentrated to 0.4 ml.

## Bacterial overlay assay

Native cationic proteins were separated by acid urea polyacrylamide gel electrophoresis (AU-PAGE) (21). Gels were either stained with PAGE-Blue (Fermentas, St. Leon-Rot, Germany) or washed  $3 \times 8$  min in 10 mM phosphate buffer (pH 7.0) for a bacterial overlay assay. After washing, the gel was incubated for 3 h on *B. subtilis*-inoculated nutrient-poor agarose (see Agar Diffusion Assay). After removal of the gel, the agarose was overlaid with double-strength nutrient agarose and treated as described for the agar diffusion assay.

#### **Immunoblotting**

Proteins were separated by tris-tricine SDS-PAGE, as described previously (22), and transferred onto nitrocellulose membranes (Schleicher and Schuell, Keene, NH, USA). Membranes were subsequently blocked with 5% nonfat dry milk (Bio-Rad, Veenendaal, The Netherlands) plus 0.5 M NaCl and 0.5% (v/v) Tween-20 in 10 mM Tris-HCl, pH 7.5 (rinse buffer), for 1 h. Blocked membranes were incubated with affinity-purified antibee defensin-1 antibody at 1.4

 $\mu g/ml$  in rinse buffer for 2 h. After incubation with primary antibody, membranes were washed  $2\times$  for 15 min in rinse buffer, incubated with horseradish peroxidase-labeled goatanti-rabbit secondary antibody (Jackson ImmunoResearch West Grove, PA, USA) at 0.4  $\mu g/ml$  in rinse buffer for 1 h, and washed again for 10 min. in rinse buffer and 5 min in PBS, respectively. The membrane was developed using a DAB liquid substrate kit (Sigma).

## Purification of antibacterial peptide from honey

An amount of >5-kDa honey retentate equivalent to 13 ml of honey was dissolved in loading buffer (3M urea in 5% acetic acid with methyl green as tracking dye) and loaded on a preparative acid-urea PAGE, as described previously (21) with slight modifications. A cylindrical gel (3.7 cm diameter, 6 cm height) in a model 491 Prep Cell (Bio-Rad) was prepared, prerun at reversed polarity for 3 h at 150 V in 5% acetic acid at 4°C, and protein was electrophoresed at 40 mA with reversed polarity. Protein was eluted in 5% acetic acid at 0.5 ml/min and collected in fractions of 2 ml. Fractions were assessed for protein composition by tris-tricine SDS-PAGE and for antibacterial activity by bacterial overlay assay. Fractions containing purified antibacterial protein were pooled, concentrated, dialyzed against 0.01% acetic acid in a 3.5-kDa molecular weight cutoff MINI Slide-A-Lyzer tube (Pierce, Rockford, IL, USA), freeze-dried, and dissolved in demineralized water.

# Protein identification by V8 digestion with subsequent mass analysis

Duplicate fractions (estimated to contain  $\sim$ 2 µg of protein each) were adjusted to 50 mM sodium phosphate (pH 7.9)

and 5% (v/v) acetonitrile. Approximately 0.5 µg of endoproteinase Glu-C (Fluka) was added per fraction and incubated at 25°C overnight. The resulting peptide mixtures were purified and concentrated with the aid of C18 ziptips (Millipore) and eluted in 10 µl 90% (v/v) acetonitrile and 1% (v/v) formic acid. The samples were checked for the presence of nonautodigest peptides with a reflectron MALDI-TOF mass spectrometer (MALDI; Waters, Milford, MA, USA). Next, samples were analyzed with ESI-tandem mass spectrometry (MS/MS). Data were acquired with a QT of 1 (Waters) coupled to an Ultimate nano-LC system (LC Packings Dionex, Sunnyvale, CA, USA). One microliter of peptide mixture was diluted in 10 µl of 0.1% TFA. The peptides of both samples were separated on a nanoanalytical column (75 µm i.d. × 15 cm C18 PepMap; LC Packings Dionex) using a standard gradient of acetonitrile in 0.1% formic acid. The flow of 300 nl/min was directly electrosprayed in the QT of 1 operating in data-dependent MS and MS/MS mode. The resulting MS/MS spectra were analyzed with Mascot software (Matrix Science, Boston, MA, USA). In both fractions, a doubly charged ion (VTCDLLSFKGQVND, mass 1537.8) with a sequence corresponding to the mature N terminus of bee defensin-1 could be identified (MOWSE scores >73).

#### RESULTS

Hydrogen peroxide is produced by the *Apis mellifera* (honeybee) glucose oxidase enzyme on dilution of honey. RS honey diluted to 40 to 20% accumulated high levels of  $\rm H_2O_2$  24 h after dilution, with a maximum of  $\rm 5.62\pm0.54~mM~H_2O_2$  formed in 30% honey (**Fig. 1***A*). The addition of catalase reduced  $\rm H_2O_2$  to negligible

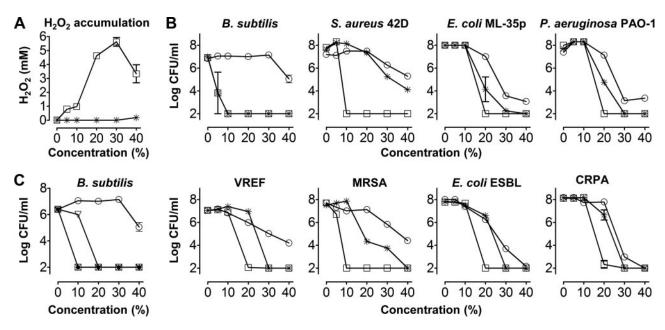


Figure 1. Contribution of  $H_2O_2$ , sugars, and MGO to the bactericidal activity of honey after 24 h. A) Mean  $\pm$  sE hydrogen peroxide accumulation in different concentrations of honey, without catalase (squares) or with catalase added (asterisks). B) Bactericidal activity against indicated laboratory strains (top row) and against clinical isolates of vancomycin-resistant E. faecium (VREF), methicillin-resistant S. aureus (MRSA), extended-spectrum  $\beta$ -lactamase-producing E. coli (E. coli ESBL), and ciprofloxacin-resistant P. aeruginosa (CRPA) (bottom row). Bacteria were exposed to various concentrations of honey (squares), honey with catalase added (asterisks), or to honey-equivalent sugar solutions (circles). C) Killing of B. subtilis by honey in incubation buffer without addition (squares), with catalase (asterisk), with glyoxalase (small solid circles), or with catalase and glyoxalase I (inverted triangles), added to neutralize  $H_2O_2$  and MGO, respectively, or by a honey-equivalent sugar solution (circles). Data are mean  $\pm$  se log-transformed bacterial concentration (CFU/ml).

levels (Fig. 1A) and markedly reduced the bactericidal activity against all bacteria tested, except B. subtilis (Fig. 1B). However,  $\rm H_2O_2$ -neutralized honey exerted stronger bactericidal activity than equivalent sugar solutions (Fig. 1B). This indicates that  $\rm H_2O_2$  is important for the bactericidal activity of honey, but that additional factors must also be present. As B. subtilis was the most susceptible bacterium for nonperoxide bactericidal activity, we used it for identification of additional bactericidal factors.

The honey bactericidal compound MGO can be converted into S-lactoylglutathione (SLG) by glyoxalase I, and this product can be measured spectrophotometrically. RS honey contained 0.25  $\pm$  0.01 mM MGO. We aimed to apply glyoxalase I to neutralize the bactericidal activity of MGO in honey. This required that SLG, the reaction product of MGO, would be nonbactericidal. Indeed, the activity of up to 20 mM MGO was neutralized by conversion into SLG (Supplemental Fig. 1), indicating that SLG up to high concentrations did not kill the bacteria. Neutralization of MGO or H<sub>2</sub>O<sub>2</sub> alone did not alter bactericidal activity of RS honey, but simultaneous neutralization of MGO and H<sub>2</sub>O<sub>2</sub> in 10% honey reduced the killing of B. subtilis by 4-logs (Fig. 1C). At higher concentrations of honey, the bactericidal activity was not affected by neutralization of H<sub>2</sub>O<sub>2</sub> and MGO (Fig. 1C), indicating that still more factors were involved.

As a first step to characterize the unknown bactericidal factors, we size-fractionated honey by ultrafiltration with a 5-kDa molecular weight cutoff membrane. Unfractionated honey produced a small zone of complete bacterial growth inhibition and a larger zone with partial growth inhibition in an agar diffusion assay with

207

114

78

53

36

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19

7

M ret. M ret.

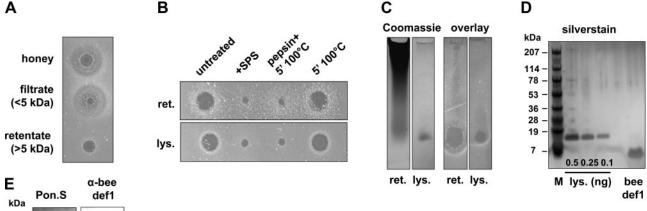
*B. subtilis* (**Fig. 2***A*). After ultrafiltration, the factors that caused complete and partial bacterial growth inhibition were separated and were present in the >5-kDa retentate and the <5-kDa filtrate, respectively (Fig. 2*A*).

Ion exchange chromatography of the retentate indicated a cationic nature of the antibacterial factors. Indeed, the polyanionic compound SPS abolished the antibacterial activity of the retentate (Fig. 2*B*). Moreover, pepsin treatment also abolished this activity (Fig. 2*B*). Together, this implies that cationic antibacterial proteins were present.

We separated cationic proteins in the retentate using a native acid-urea PAGE gel, and allowed the separated components to diffuse from this gel into a *B. subtilis*-inoculated agar to identify antibacterial proteins. This yielded a single zone of bacterial growth inhibition that corresponded to a protein band in a Coomassie-stained gel run in parallel (Fig. 2C). This protein was purified from a larger amount of retentate using preparative acid-urea PAGE (Fig. 2D), and identified by peptide mass analysis as bee defensin-1.

To specifically assess the contribution of bee defensin-1 to the bactericidal activity of honey, an antibee defensin-1 antibody was raised (Fig. 2E). Like SPS, this antibody negated all bactericidal activity of the >5-kDa retentate against B. subtilis (Fig. 3A). The <5-kDa filtrate had only minor bactericidal activity (Fig. 3A), but this was not due to cationic compounds, since SPS failed to neutralize this activity (Fig. 3A). Thus, bee defensin-1 was the only cationic bactericidal compound present in RS honey.

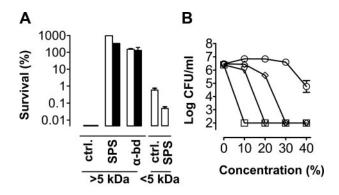
Next, we assessed the contribution of bee defensin-1 to the bactericidal activity of nonfractionated honey



**Figure 2.** Identification of bee defensin-1 in honey. *A*) Honey was fractionated by ultrafiltration using a 5-kDa molecular weight cutoff filter tube; antibacterial activity of 2.5  $\mu$ l of 80% honey, and equivalent amounts of the <5-kDa filtrate and >5-kDa retentate, were tested in an agar diffusion assay. *B*) Retentate equivalent to 7.5  $\mu$ l of undiluted honey was tested for the presence of cationic and proteinaceous antibacterial components. Activity of cationic components was neutralized by adding SPS, and protein was digested with pepsin, followed by 5-min inactivation at 100°C. As control, incubation for 5 min at 100°C without pepsin was performed. Activity in retentate (ret.) was compared with that of 0.2  $\mu$ g hen egg white lysozyme (lys.). *C*) To identify cationic antibacterial proteins in retentate, amounts of this fraction equivalent to 750  $\mu$ l honey, and 3  $\mu$ g lysozyme as a reference, were run in duplicate sets on a single native acid-urea PAGE gel. One half of the gel was Coomassie-stained (left); other was used for a bacterial overlay assay with *B. subtilis* (right). *D*) Silverstained tris-tricine SDS-PAGE of different amounts of lysozyme and preparative

acid-urea PAGE-purified bee defensin-1, separated by an empty lane. *E*) Retentate separated on tris-tricine SDS-PAGE, blotted to nitrocellulose, stained with either Ponceau S (Pon. S, left) or immunostained with antibee defensin-1 (right).

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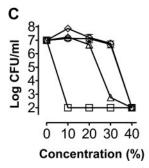


Figure 3. Roles of bee defensin-1 and pH in bactericidal activity of honey against B. subtilis. A) Contribution to bactericidal activity of cationic components in general and of bee defensin-1 specifically was tested by neutralization with SPS or with antibee defensin-1 antibody ( $\alpha$ -bd), respectively, at concentrations of retentate equivalent to 20% honey (open bars) and 40% honey (solid bars); ctrl. indicates survival without neutralization. B) To assess the contribution of bee defensin-1 to bactericidal activity of unfractionated honey, B. subtilis was incubated in various concentrations of honey in incubation buffer (squares), or with catalase and glyoxalase I added either without (triangles) or with SPS (diamonds), or in a honey-equivalent sugar solution (circles). C) To assess the contribution of the low pH to the bactericidal activity of honey, B. subtilis was incubated in various concentrations of honey in incubation buffer (squares), or with catalase, glyoxalase I, and SPS added either without (triangles) or with neutralization to pH 7 (diamonds), or in a honey-equivalent sugar solution (circles). After 24 h, numbers of surviving bacteria were determined. Data are mean ± sE log-transformed bacterial concentration (CFU/ml).

against *B. subtilis*. As previously observed,  $\geq 20\%$  honey retained bactericidal activity when  $H_2O_2$  and MGO were neutralized. Additional neutralization of bee defensin-1 strongly reduced the bactericidal activity of 20% honey but did not affect the activity of 30 and 40% honey (Fig. 3*B*). So, bee defensin-1 contributed to the bactericidal activity of honey, but still other bactericidal factors were involved.

Honey has a low pH, mainly because of the conversion of glucose into hydrogen peroxide and gluconic acid by glucose oxidase. This low pH might also contribute to the bactericidal activity of honey (23). Titration of the pH of 40-10% RS honey from 3.4-3.5 to 7.0, combined with neutralization of  $H_2O_2$ , MGO and bee defensin-1, reduced the bactericidal activity of honey to a level identical to that of a honey-equivalent sugar solution (Fig. 3C). Thus, with this experiment, we

succeeded in identifying all bactericidal factors in RS honey responsible for killing of *B. subtilis*.

The contribution of the identified bactericidal factors to activity against antibiotic-susceptible and -resistant strains of various species was tested with honey diluted to 20%, since this killed the entire inocula of all bacteria tested independent of sugar (Fig. 1). Simultaneous neutralization of  $\rm H_2O_2$ , MGO and bee defensin-1 negated all activity (**Fig. 4**), showing that these were the major factors responsible for broad spectrum bactericidal activity of honey.

We studied the contribution of the honey bactericidal factors in more detail by neutralizing the factors individually or combined. Neutralization of H<sub>2</sub>O<sub>2</sub> alone strongly reduced the bactericidal activity against all bacteria tested except B. subtilis (Fig. 4). Neutralization of MGO alone strongly reduced killing of E. coli and P. aeruginosa strains (Fig. 4). Neutralization of bee defensin-1 alone reduced killing of VREF, but not of the other bacteria tested (Fig. 4). When compared to neutralization of MGO alone, the additional neutralization of bee defensin-1 reduced killing of all bacteria tested, except E. coli ESBL (Fig. 4). In summary, H<sub>2</sub>O<sub>2</sub>, MGO, and bee defensin-1 differentially contributed to the activity of honey against specific bacteria, and their combined presence was required for the broad-spectrum activity.

#### **DISCUSSION**

All bacterial species tested were susceptible to different combinations of bactericidal factors in honey, indicating that these bacteria were killed *via* distinct mechanisms. This clearly demonstrates the importance of the multifactorial nature of honey for its potent, broadspectrum bactericidal activity.

Some factors had overlapping activity. For instance, the activity of bee defensin-1 against most bacteria was only revealed after neutralization of MGO. This clearly demonstrates the importance of neutralizing known bactericidal factors in honey to reveal the presence of additional factors. Similarly, the contribution of the low pH for activity of honey against *B. subtilis* was only revealed when H<sub>2</sub>O<sub>2</sub>, MGO, and bee defensin-1 were simultaneously neutralized.

In other situations, bactericidal activity depended on the combined presence of different factors. Thus, the activity of honey against  $E.\ coli$  and  $P.\ aeruginosa$  was markedly reduced by neutralization of either  $H_2O_2$  or MGO. Alternatively, the activity of certain bactericidal factors likely is more potent in the context of honey than as pure substances. This is most clearly illustrated by the activity of MGO. When tested in a buffer,  $\geq 0.3$  mM MGO was required for activity against  $B.\ subtilis$  (Supplemental Fig. 1). In contrast, as little as 0.05 mM MGO, the concentration in 20% RS honey, was sufficient to substantially contribute to the bactericidal activity. This suggests that the presence of the other bactericidal factors in honey enhanced the effect of

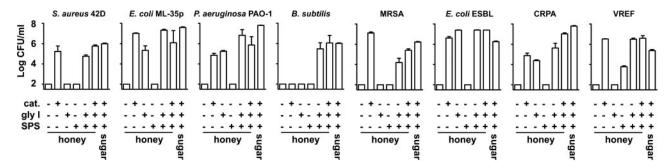


Figure 4. Effect of neutralization of  $H_2O_2$ , MGO, and bee defensin-1 on bactericidal activity of honey. Hydrogen peroxide, MGO, and bee defensin-1 were neutralized in 20% honey by adding catalase (cat.), glyoxalase I (gly I) and SPS, respectively. Bactericidal activity was tested against indicated laboratory strains (left 4 panels) and against clinical isolates of VREF, MRSA, *E. coli* ESBL, and CRPA (right 4 panels). A sugar solution equivalent to 20% honey was used as a reference. After 24 h, numbers of surviving bacteria were determined. Data are mean  $\pm$  se log-transformed bacterial concentration (CFU/ml).

MGO. It is not possible to quantify the contribution of the different factors to honey bactericidal activity since, as we have shown, these factors may have redundant activity, be mutually dependent, or have additive or synergistic activity depending on the bacterial species targeted.

We have demonstrated for the first time that honey contains an antimicrobial peptide, bee defensin-1, and that this peptide substantially contributes to the bactericidal activity. Bee defensin-1 was previously isolated from royal jelly (24), the major food source for bee queen larvae (and then referred to as "royalisin"), and was identified in honeybee hemolymph (18). Royal jelly is produced by young worker bees and contains their hypopharyngeal and mandibular gland secretions (25, 26). Bee defensin-1 mRNA has been identified in the hypopharyngeal gland of young worker bees (18), suggesting this gland is involved in production of bee defensin-1 found in royal jelly (24). When worker bees age, they become the major producers of honey. Major differences develop in morphology and protein expression of their hypopharyngeal glands (27, 28), e.g., several important carbohydrate-metabolizing enzymes, including glucose oxidase are expressed (29). The bees add the secretion from their hypopharyngeal glands to the collected nectar. The carbohydrate-metabolizing enzymes then convert sucrose to glucose and fructose, and glucose oxidase converts the glucose to hydrogen peroxide and gluconic acid. These latter compounds presumably are involved in prevention of microbial spoilage of unripe honey (11). Since we have found bee defensin-1 in honey, this suggests that after the transition in hypopharyngeal gland function of the worker bees with age, the gland still produces bee defensin-1. This peptide, therefore, likely contributes to protection of both royal jelly and honey against microbial spoilage.

It remains to be established whether bee defensin-1 is also present in other honeys. In Manuka honey, no evidence was found for the presence of antimicrobial peptides (30). For several other honeys, proteins were reported to contribute to the antibacterial activity (31, 32), but their identity remains unknown. Using our antibee defensin-1 antibody, we aim to assess the role of

bee defensin-1 for the antibacterial activity of other honeys.

Previous studies regarding the effect of low pH to antibacterial activity of honey have yielded conflicting results (11). In our study, the contribution of the low pH for activity against *B. subtilis* was only revealed on inactivation of all other bactericidal factors. So, in other studies, which did not employ an approach of neutralization of bactericidal factors in honey, the contribution of the low pH of honey may easily have been overlooked.

Much effort has been put into identification of phenolic antibacterial components in honey (11). Several of these compounds have been isolated from honey, but as they were tested at concentrations far exceeding those in honey, no conclusions can be drawn regarding their contribution to honey bactericidal activity (11). Our data do not show a role of phenolic compounds in RS honey bactericidal activity.

Our approach of selectively neutralizing individual bactericidal factors present in a medical-grade honey allowed us to unravel the multifactorial bactericidal activity of a honey for the first time. We presently use the same approach to assess the contribution of these factors to activity of other honeys, and simultaneously to screen for novel bactericidal factors. Such honeys, or isolated components thereof, may serve as novel agents to prevent or treat infections, in particular those caused by antibiotic-resistant bacteria.

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